NZ POOL INDUSTRY ASSOCIATION DIPLOMA COURSE 2021 TEST PAPER

Dip.Pool Tech

New Zealand Pool Industry Association Inc, PO Box 170769 Greenlane 1546

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PAPER	7 TEST Your Name			
THE CHEMICAL BALANCE OF CONCRETE POOL WATER (Background Information)				
Since it is	rare to find water which is in perfect equilibrium with the pool's interior,			
	1. WHATis required to correct the imbalance in the solids content of the water.			
	We have to bring the water to the point of WHAT 2 by adding certain substances in the right proportion, rather than wait until this state is achieved naturally.			
	3means the adding of specified quantities of chemicals to the pool water to achieve equilibrium between the pool water and the internal (often plaster) surface of the pool.			
	The correct chemical balance is very important as it has an effect on:			
5				
7				
	The chemical balance of the pool water is the combined effect of:			
8				
9	· Becommon to the control of the con			
10				
	na visika na uma na mana ana p ure visingerete desegniar synthest, 57 (1. 1747,			
12				
	Why is the pH of the pool water important?			
	13.			
	SATURATION INDEX (SI)			
	What does this Index measure?			
14				
	Why is it important?			

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	Why is it not used as much as in the past? 16	
It was po	ointed out that one of the roles of the chemical treatment is to make the water of and equipment. Also it was mentioned that different pool finishes require different	ompatible with the pool's finishent treatments.
Please e	expand on the above statement	
	17.	
F31 =		
	so stated that the water which is out of chemical balance etched the plaster fine	sh finish or it formed a deposit
	WHAT happens? 18.	
to etch a	se of the vinyl lined pool, however, such etching is impossible by the unsatural and the vinyl is insoluble in the water. Also it is difficult to form a hard scale on the the third words, there was:	ed water as there is no plaster the smooth surface of the vinyl
	19 between the water and the vinyl linir	ıg.
In the ca	se of the over-saturation the excess is not held by the water, and it is likely to be	pecome
	20 and 21	m specima agenti. I
This situ	ation may also apply to moulded fiberglass pools.	
	Pool finishes can be classified into two categories:	
22	FINISHES like marblesheen, exposed aggregate, the grout	ing around the tiles, slates etc.
23	FINISHES like vinyl lining, paint, fibreglass, any that does n	ot contain cement or plaster.
be treated	If there are a couple of rows of tiles at the water line of a fibreglass pool with cement-based grouting as if the pool were fully tiled i.e. the pool's interior should be judged by its weakest point (the ground water may cause damage.	ng between the tiles, the water should ling, not the tiles) where the
	How do you lower the calcium hardness of pool water?	
water wi	This is a difficult process, and it should be attempted if there are no alternation the softer water.	ves, like replacing part of the
Procedu	re:	
24.	* .	
25.		
26.		
27.		1100
28.		
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TOTAL A	ALKALINITY	
		rease the level of total alkalinity and it is recommended that this is used when hieve the desirable saturation index. The total alkalinity must be established
	Is Soda ash suitable to use	in this instance YES NO.
		to the concentration of the sodium bicarbonate. The higher the total alkalinity ore stable when the total alkalinity is above 100 mg/L.
The appr	oximate pH values –vs - total	alkalinity are listed below:
Total Alkalinity - mg/L		рН
	20 50 100 <u>125</u> 160 200 240	7.0 7.4 7.7 7.8 7.6 – 7.8 ideal for most residential pools 7.9 8.0 8.2
	How do you increase the tota	l alkalinity of the pool water?
 29		should be used
30 31 E	How do you lower the total al	kalinity of the pool water?
, p	NOTE: After adding the above Note: After adding the above within a few	products to the water the pH will be lower than expected, however, the final days.
рН		
	cannot be altered without che of the water should be maint	ndant on the amounts of acidic or alkaline chemicals present in the water. It anging the acidity or the total alkalinity. Irrespective of the pool's finish the pH ained above 7 at all times. The only exception to this rule is in the case of otential damage by osmosis, may develop unsightly black spots. The 6.8.
What may	y occur when the pH of the poo	ol water is below 7:

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37	
When the pH of the pool water is 7.0.	
38 the pH and ev cause significant changes.	ven small amounts of acids or alkaline materials
39 the of the	water changes with the pH.
When the pH of the pool water is between 7.2 and 8.0	
40 this level isIS or IS NOT the recor	mmended pH range for all pools;
41 as the total alkalinity increases the pH becon	nes
42 chlorine and bromine sanitisers are	<u> </u>
43 people can stay in the water for long periods without	The second secon
When the pH of the pool water is above 8.0.	
44 is ineffective with the increasing pH and much to maintain efficient sanitisation;	n higher levels are required
45 the likelihood of "milkiness" and	_increases especially above pH 8.4;
46 the water depletes the skin of	_especially in the warmer waters.
When the pH has to be altered it is carried out by increasin	g or decreasing the
TOTAL DISSOLVED SOLIDS	and they are he reduced only by
As it was mentioned earlier the dissolved solids cannot be filtered out	aging et grant significant de la reta
48 and 49	
RECOMMENDED STARTING-UP PROCEDURES	
and a storage for the state of	

- end -